



# **Advancing the Art of Winemaking through/with Science: Impact of Grape Ripening on Wine Phenolics and Sensory Attributes**

**Jim Harbertson**

**Washington State University**

# Introduction

- **Grape Ripening and its Influence on Wine Composition (DOA)**
- **Fruit and wine relationships are complex**
- **Not many relationships between grapes and wine are:**
- **WYSIWYG “wiz-ee-wig”**

# Complex Relationships: Grape and Wine Flavor

- **What you taste in fruit is not what you taste in wine?**
  - **Some flavor compounds are bound as precursors**
  - **Some flavor compounds are not**
- **Flavor compounds are changing during ripening**
- **Some influenced by vineyard practices**
  - **Sun exposure: IBMP, TDN**
- **How does wine composition influence this relationship?**
  - **Ethanol solvates hydrophobic compounds that we smell and taste**

# Complex Relationships: Color and Tannins

- **Relationship between fruit and wine tannin is awful (DOA)**
  - Cell wall compounds: polysaccharide
  - Sponge of complex polysaccharides that must be satisfied (bound) before you can get free tannins into wine
  - Ethanol has some influence on extraction
- **Relationship between fruit and wine anthocyanins isn't great either (RBB)**
  - Copigmentation: Influence of [A] & Co-factors
  - Extraction equilibrium (adsorption/desorption phenomenon)
  - Not quite like the sponge but similar
  - Ethanol has no influence on extraction

# How does fruit composition influence polymeric pigment formation?

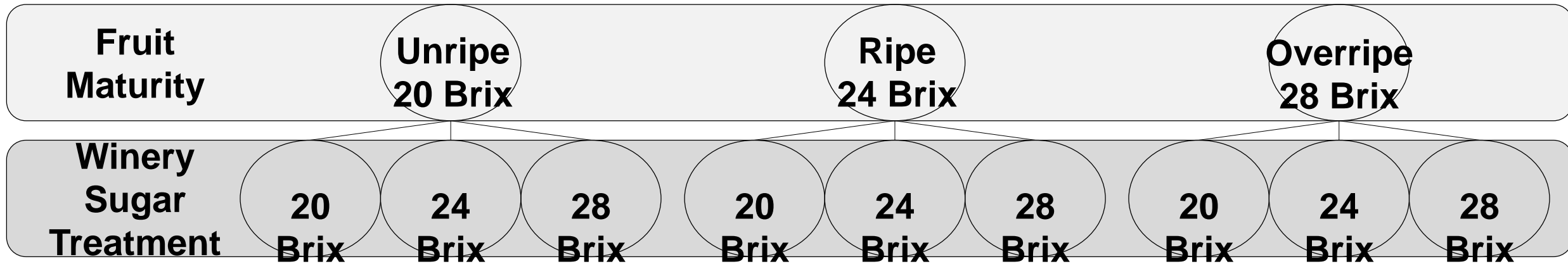
- **Chemical Train Wreck: Polymeric Pigments (DOA)**
  - Reaction between an anthocyanin and variety of wine components
  - Primarily tannins
  - Form stable color in wines
  - Coloration is less effected by pH changes and bisulfite bleaching
  - Modify mouth feel over time by decreasing astringency (theoretical still)
- **Anthocyanin:Tannin thought to drive formation**
  - Evidence in literature is negligible
  - Heat, O<sub>2</sub>, Lack of O<sub>2</sub>: all influence formation
  - Reactions take time so we will need a way around this

# Experimental Design

- **Pick fruit at different soluble solids: 20 Brix, 24 Brix, 28 Brix**
- **Represent different winemaking eras and extraction effect**
  - ~ 12 %, 14%, & 16% (v/v) Ethanol
- **Wines not skin or seed extracts**
  - **Phenolics extracted from wines Day 10 and subjected to heat treatments**
    - Heat treatment based upon work done by Vidal et al. 2002
- **Ethanol is controlled for at each harvest by dilution or sugar addition**
- **Cultivars that naturally have different A:T ratios selected for study**
  - **Syrah (High Pigment: Mid Tannin)**
  - **Cabernet Sauvignon (High Pigment: High Tannin)**

# Winemaking Procedure

- Syrah and Cabernet Sauvignon
- Wines replicated sugar content of other maturity treatments
  - Controlled for maturity vs. ethanol effects



- Experiment designed so wines would have a range of anthocyanin, tannin, and A:T

# Winemaking Procedure

- **Wines fermented in triplicate**
  - 200 L scale, 54 total wines
  - TJ/Boulton Fermentors
- **Inoculated with EC 1118 ( $10^6$  cells/mL)**
- **Simultaneous ML fermentation (~48 hours post using VP41)**
- **Nutrient Addition**
  - FermaidK (0.25 g/L), DAP (200ppm), GoFerm (0.3g/L)
- **No acidity adjustments**
  - Water for saignée/water back had 5 g/L tartaric acid
- **Chaptalization with 80 Brix sugar solution**
- **10 day maceration**





# Aging

- **Wines were collected at day 10 of fermentation**
  - Remaining sugars and organic acids were removed (XAD7)
  - Dissolved in same volume model wine (14% alcohol, 5 g/L TA, pH=3.5)
- **Aged at 30°C for 4 months**
  - Samples collected once a month
  - Cellar aged samples collected 6 months after fermentation
- **Analysis of polymeric pigment, anthocyanin, tannin, and total phenolics performed by protein precipitation,  $\text{HSO}_3^-$  bleaching assays and  $\text{FeCl}_3$**
- **HPLC methodologies also done but not shown today**

# 2015 Harvest Data

	Harvest (Pick Date)	Brix at Harvest	pH	TA (g/L)	Berry Weight (g)	Anthocyanin (mg/berry)
Cabernet Sauvigno n	Unripe (DOY 233)	19.2 c	3.41 c	9.29 a	0.82 b	0.71 c
	Ripe (DOY 260)	25.1 b	3.72 b	7.23 b	1.02 a	0.91 a
	Overripe (DOY 289)	27.5 a	3.89 a	6.91 c	0.83 b	0.82 b
Syrah	Unripe (DOY 231)	20.0 c	3.47 c	7.95 a	1.37 a	0.90 c
	Ripe (DOY 252)	24.5 b	3.73 b	8.07 a	1.36 a	1.52 a
	Overripe (DOY 286)	27.9 a	4.01 a	4.72 b	1.12 b	1.14 b

- ~ 3-4 weeks between pick dates

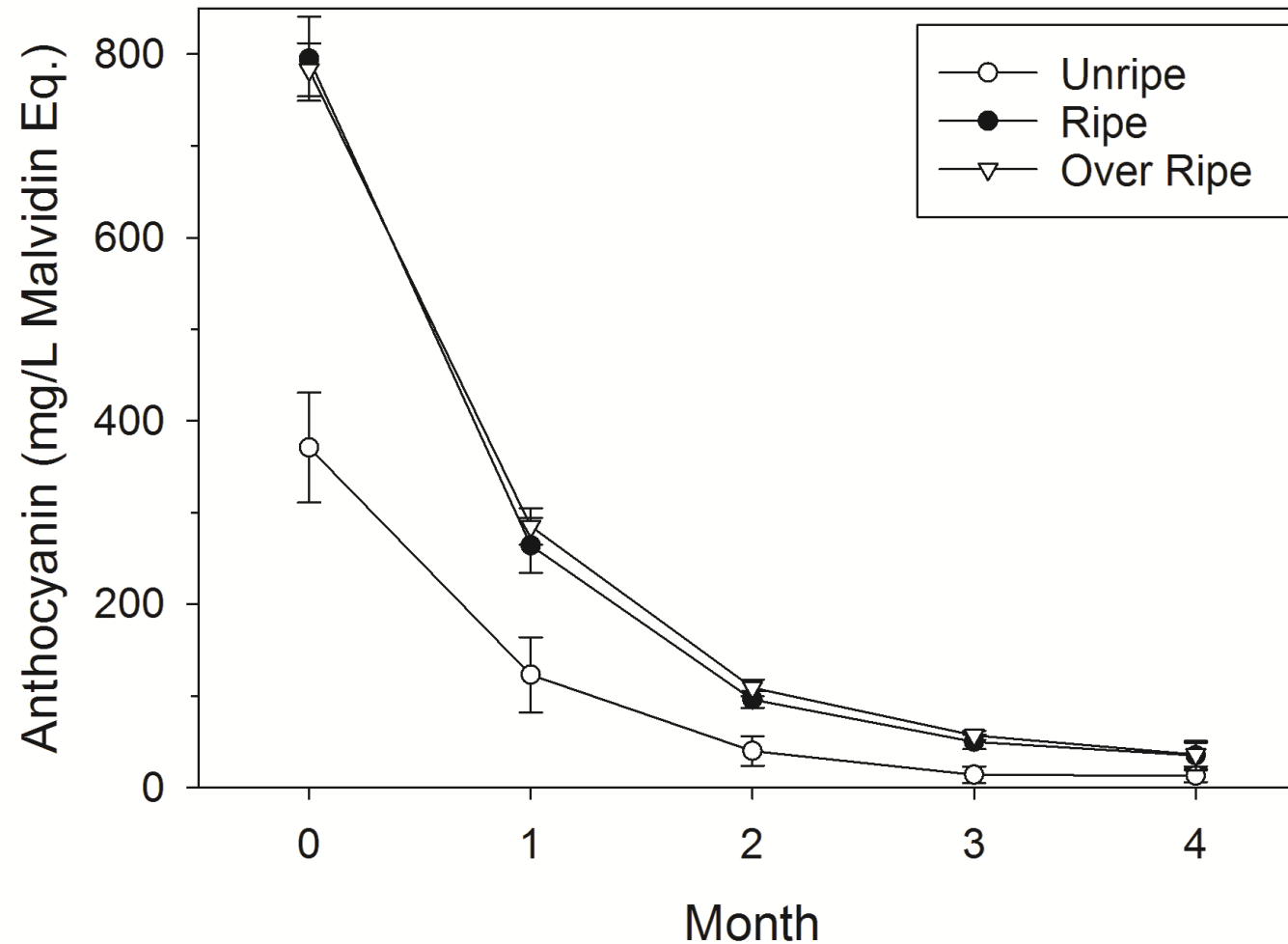
# Cabernet Sauvignon Initial Wine Phenolic Data

<b>Harvest Date</b>	<b>Anthocyanin (mg/L)</b>	<b>Tannin (mg/L CE)</b>	<b>Ratio A:T</b>
<b>Unripe DOY 233</b>	<b>371 a</b>	<b>1072 a</b>	<b>0.36 a</b>
<b>Ripe DOY 260</b>	<b>795 b</b>	<b>886 b</b>	<b>0.93 b</b>
<b>Overripe DOY 289</b>	<b>783 b</b>	<b>892 b</b>	<b>0.86 b</b>
<b>Alcohol Treatment</b>			
<b>Low</b>	<b>641</b>	<b>816 a</b>	<b>0.83</b>
<b>Medium</b>	<b>654</b>	<b>946 a</b>	<b>0.71</b>
<b>High</b>	<b>623</b>	<b>1189 b</b>	<b>0.61</b>

# Syrah Initial Wine Data

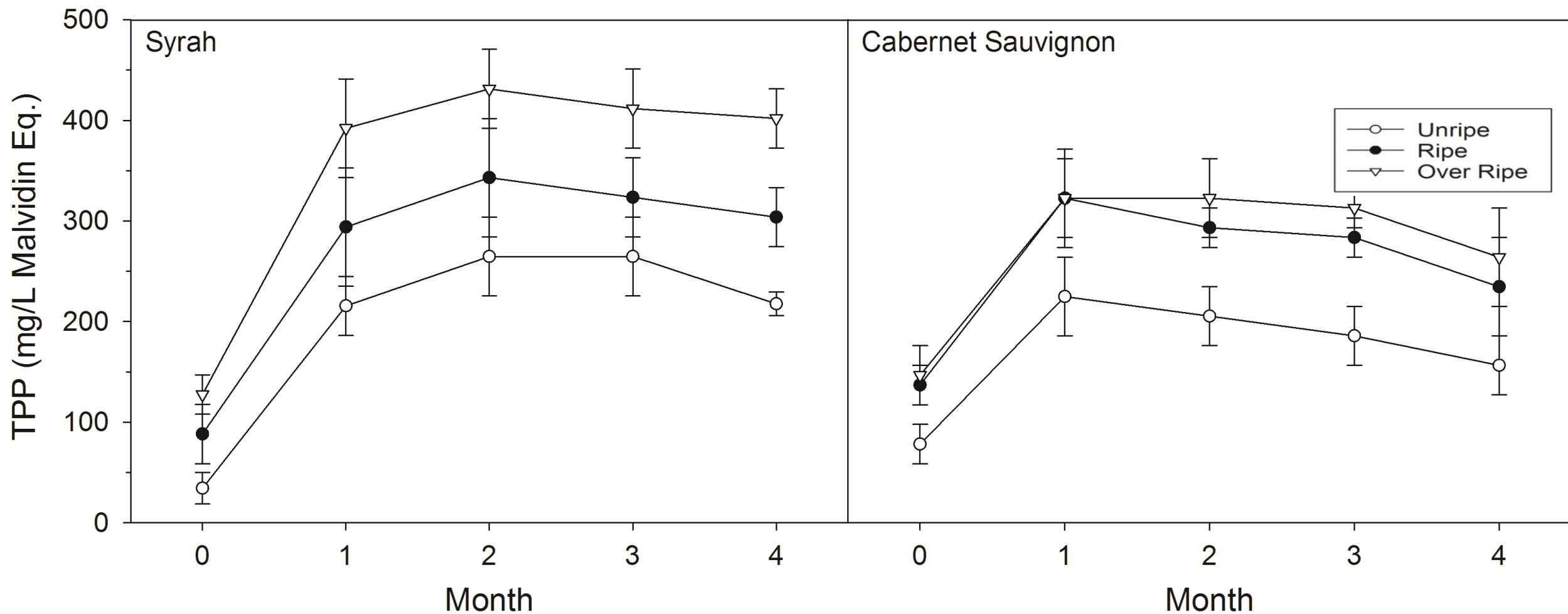
<b>Harvest Date</b>	<b>Anthocyanin (mg/L)</b>	<b>Tannin (mg/L CE)</b>	<b>Ratio A:T</b>
<b>Unripe DOY 231</b>	<b>458 c</b>	<b>374 ab</b>	<b>1.3 b</b>
<b>Ripe DOY 252</b>	<b>726 b</b>	<b>351 a</b>	<b>2.1 a</b>
<b>Overripe DOY 286</b>	<b>832 a</b>	<b>429 b</b>	<b>2.0 a</b>
<b>Alcohol Treatment</b>			
<b>Low</b>	<b>640</b>	<b>302 b</b>	<b>2.1 a</b>
<b>Medium</b>	<b>700</b>	<b>416 a</b>	<b>1.7 b</b>
<b>High</b>	<b>680</b>	<b>437 a</b>	<b>1.6 b</b>

# Anthocyanin Changes Over Time



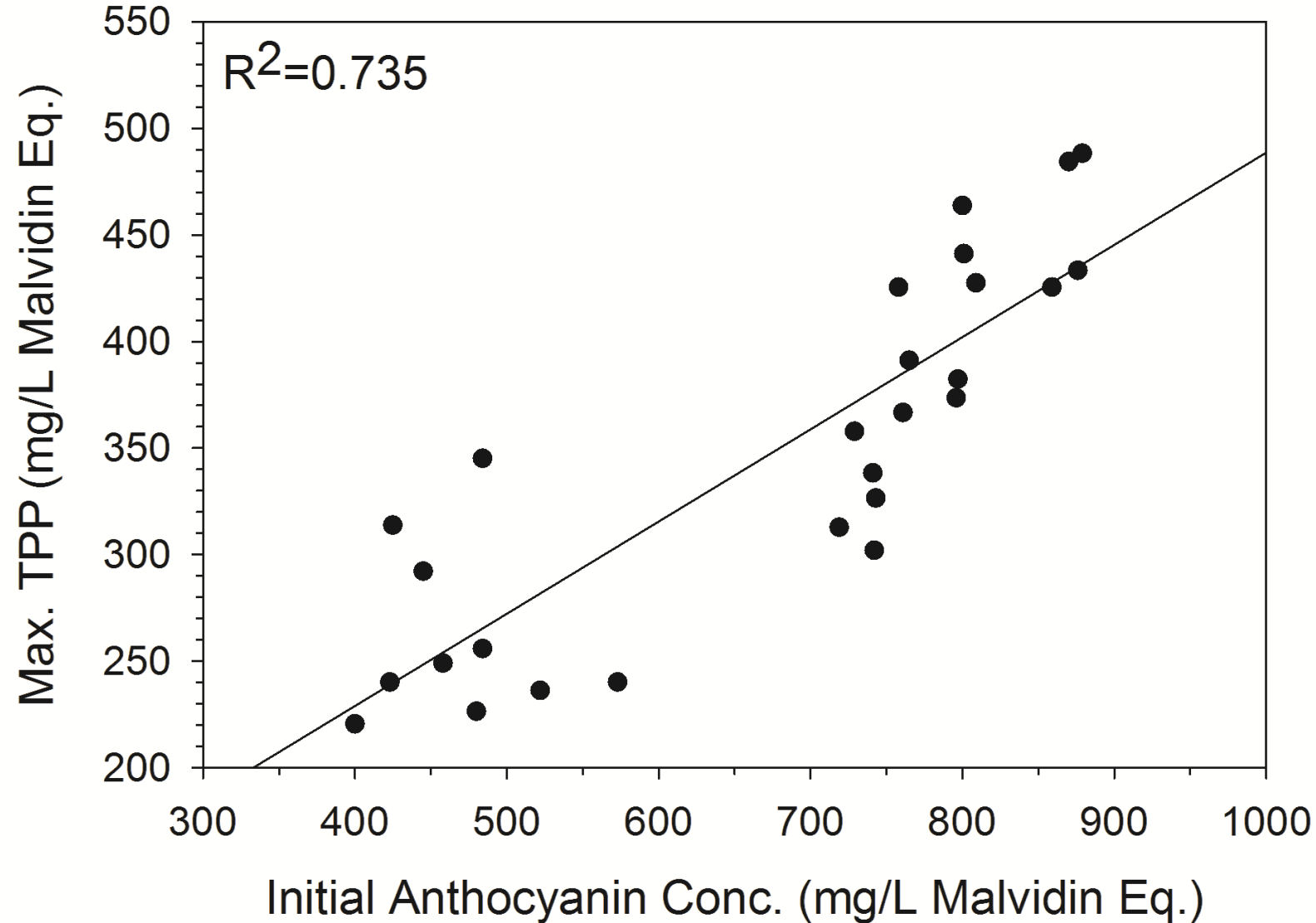
- Independent of alcohol treatment Exponential decay  $R^2$ : 0.94-0.99
- 1 month incubator=1 year cellar

# Polymeric Pigment Over Time



- **1 month incubator=1 year cellar**

# Predicting Polymeric Pigment Content (SY)



# R<sup>2</sup> Values for Other Predictors

- **Syrah**
  - A:T=0.042
  - [Tannin]=0.392
  - [Anthocyanin]=0.735
  - [Tannin] + [Anthocyanin]=0.859
- **Cabernet Sauvignon**
  - A:T=0.405
  - [Tannin]=0.02
  - [Anthocyanin]=0.670
  - [Tannin] + [Anthocyanin]=0.767



# Conclusions

- **Initial wine (not necessarily fruit) anthocyanin concentration best single predictor of long-term polymeric pigment**
  - **Higher initial anthocyanin (and tannin), higher polymeric pigment**
    - **More stable color and mouth feel modification over time**
- **Polymeric pigment formation occurs relatively rapidly**
  - **At equilibrium between formation and sedimentation after 1 month (1 year cellar)**

# Hang Time Experiment: Merlot Flavor

## Harvest 1: Unripe

20.7 ± 0.5 Brix

Chaptalize to 24 Brix

5 September 2013

*~21 Days*

Brix

## Harvest 2: Ripe

Brix

24.0 ± 0.2 Brix

Brix)

*~37 Days*

26 September 2013

## Harvest 3: Overripe

20 Brix

27.4 ± 0.4 Brix

to 24 Brix

2 November 2013

**Low** : Control (~20 Brix)

**Medium**:

**High**: Chaptalize to 28

**Low**: Saignée – +H<sub>2</sub>O to 20

**Medium**: Control (~24

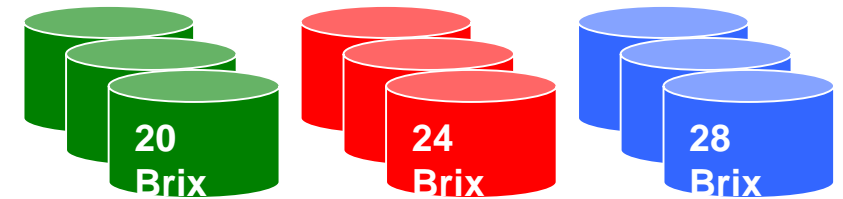
**High**: Chaptalize to 28 Brix

**Low**: Saignée- +H<sub>2</sub>O to

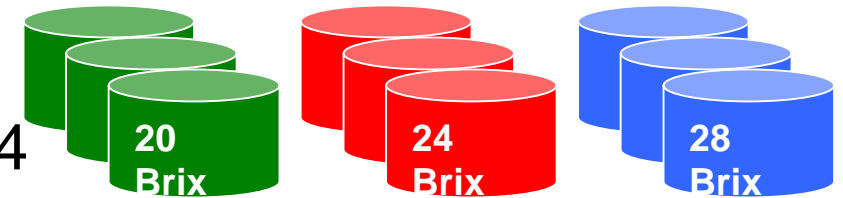
**Medium**: Saignée- +H<sub>2</sub>O

**High**: Control (~28 Brix)

UNRIPE



RIPE



OVERRIPE



# WINEMAKING

- 300 kg/Replicate
- 300 L Stainless Steel Tank
  - Treatment Replicates: n=3
- Yeast (EC-1118)
- 48 hrs. ML (VP41)
- 10 Days Contact Time
  - (26 ± 2°C)



# Fruit Data

Harvest	Brix	pH	TA (g/L)	Berry Weight	Color (mg/g FW)	Skin Tannins (mg/g FW)	Seed Tannins (mg/g FW)
UNRIPE	20.7 a	3.57 a	7.83 c	0.98 a	0.65 a	0.60 a	3.68 b
RIPE	23.9 b	3.73 b	5.56 a	1.18 b	0.73 a	0.60 a	3.06 a
Overripe dehydration	27.4 c	3.73 b	6.60 b	0.99 a	0.99 b	0.86 b	3.66 b

Overripe fruit characterized by concentration effects from dehydration

Intuitive Impacts: More color and skin tannins

Counter Intuitive Impacts: TA increase, Seed Tannin Increase

Drop in yield about 20-25% when ripening to 28 Brix

Harvest	EtOH % (v/v)	pH	TA (g/L)	RS (g/L)	Dynamic Viscosity (cP)	Density (g/cm <sup>3</sup> )
UNRIPE	13.86	3.63 a	5.01 b	3.11 a	1.35 c	0.9857 a
RIPE	14.03	3.73 b	4.52 a	2.56 a	1.29 a	0.9860 a
OVERRIPE	13.95	3.73 b	5.15 b	4.11 a	1.32 b	0.9872 b
Ethanol						
Low	11.59	3.60 a	4.86	1.94 a	1.22 a	0.9884 c
Med	14.03	3.73 b	4.88	1.89 a	1.33 b	0.9860 b

**OVERRIPE: Greater Viscosity and Lower Density**  
**More EtOH: Yeast struggle (Higher RS); Greater Viscosity, Lower Density**

Harvest	Anthos (mg/L)	SPP (A <sub>520nm</sub> )	LPP (A <sub>520nm</sub> )	Tannins (mg/L)	Total Iron Reactive Phenolics (mg/L)
UNRIPE	249 a	0.90 b	0.54 c	564 b	1571 a
RIPE	469 b	1.11 c	0.31 a	440 a	1521 a
OVERRIPE	524 c	0.82 a	0.40 b	792 c	2338 b
Ethanol					
Low	430	0.87 a	0.32 a	537 a	1655 a
Med	410	0.91 a	0.41 b	591 b	1766 b
High	403	1.06 b	0.52 c	669 c	2008 c

@ 60  
Total

DE:

**High Ethanol Impacted: Tannins, Total IRP**

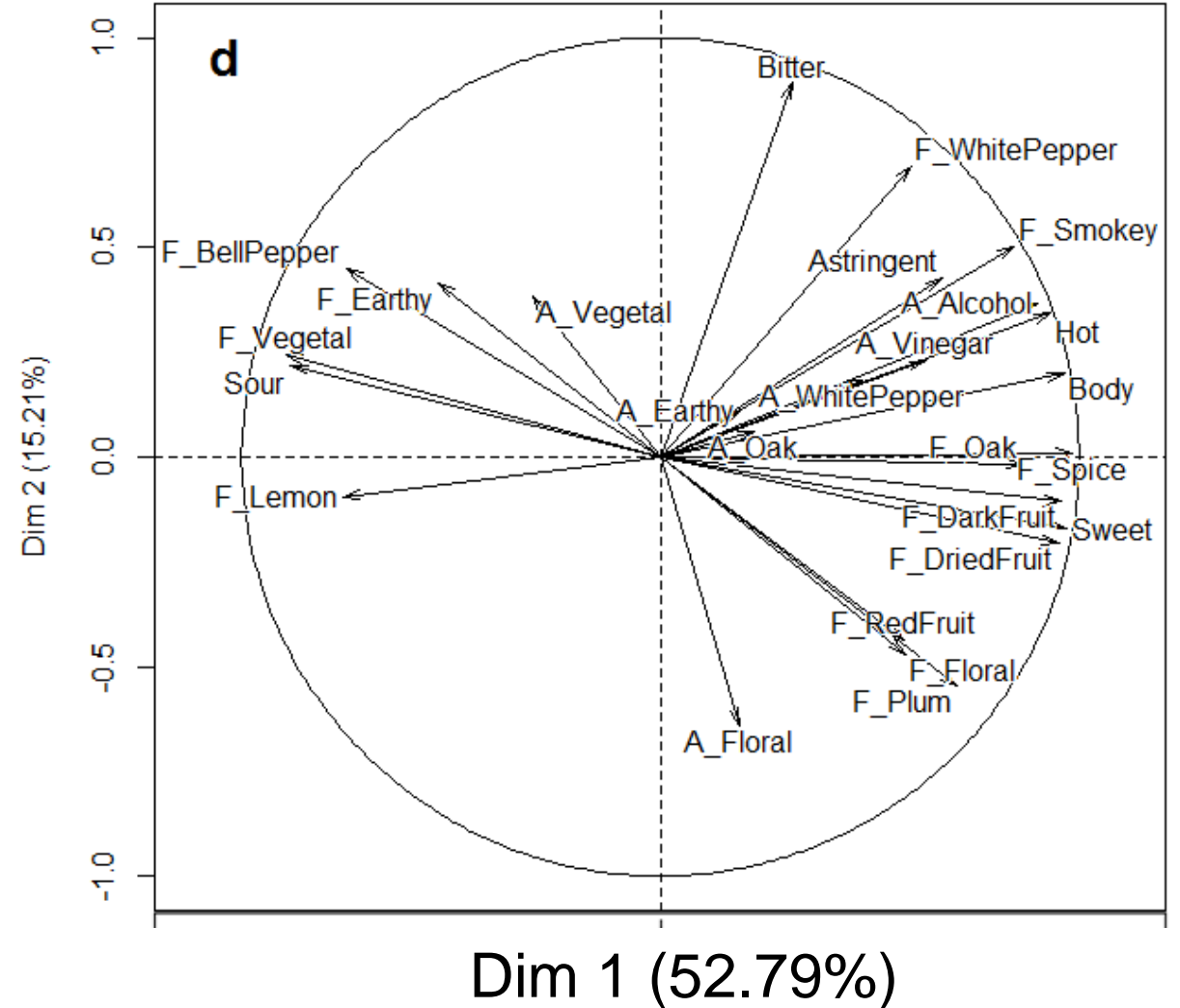
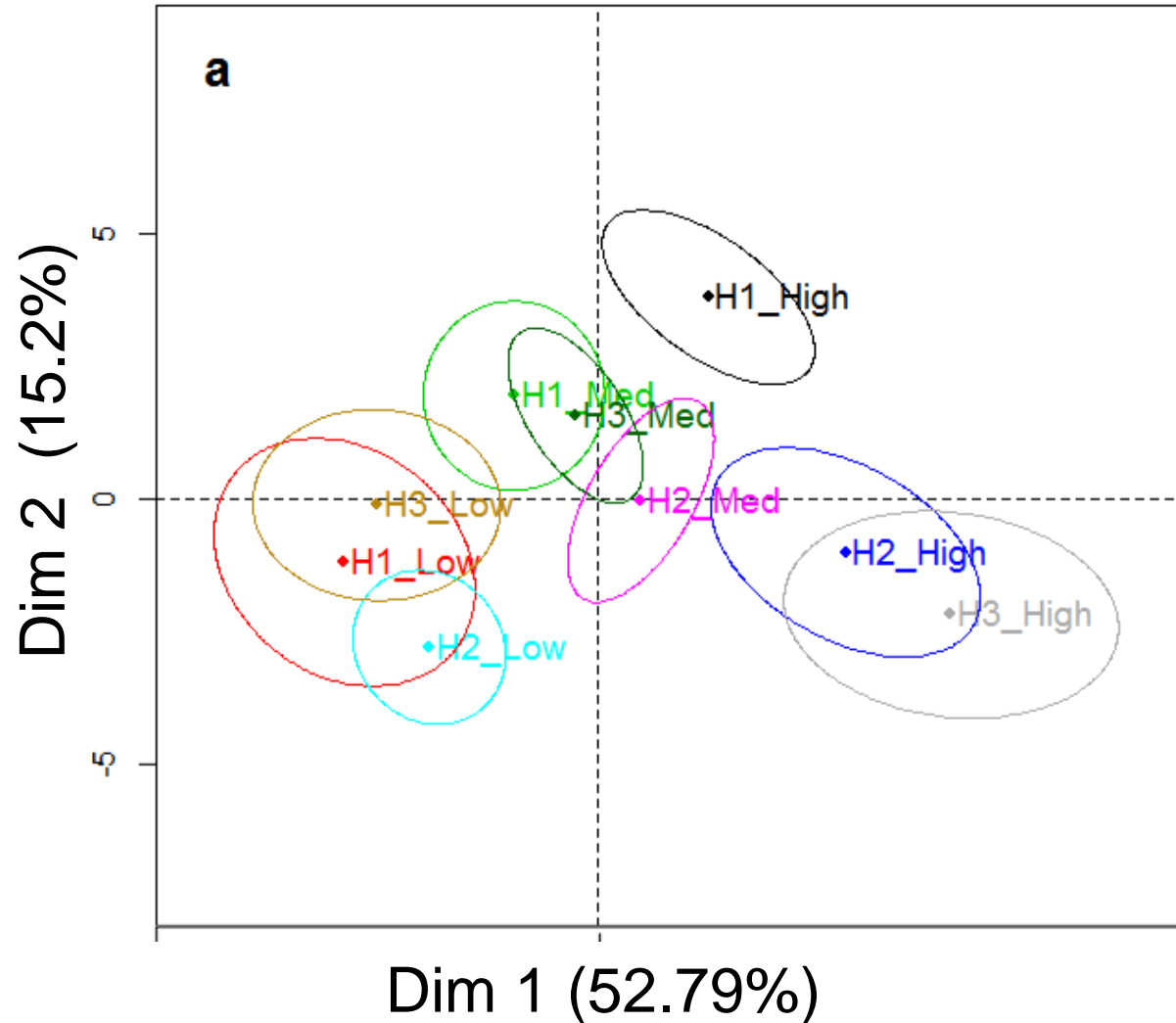
# **Sensory Panel Work**

- **Descriptive Analysis: UC Davis Sherman & Heymann**
- **Sourness, Bitterness, Astringency, Sweet, Body**
- **Aromas by aroma and flavor**
  - **Vegetal, bell pepper, smokey, white pepper, floral, spice, red fruit, plum, dried fruit, oak**

# Sensory Evaluation PCA

## Dim 1 x Dim 2

SEPARATION BY ETHANOL  
HIGH, MEDIUM, LOW





# **Ethanol Dominated Sensorial Evaluation**

- **Wines with similar ethanol were more similar to each other than the wines made from fruit picked at 20, 24 and ~28 Brix**
- **Low Ethanol: sourness, vegetal, bell pepper and earthy flavors.**
- **Medium Ethanol: vegetal, earthy and floral aromas.**
- **High Ethanol: Astringent, Bitter, Hot, Body, Sweet, Alcohol; Aromas & Flavors: Red Fruit, Plum, Oak, Smokey, White Pepper**

**Explain that one to me again**



# Henry's Law

- **Tendency of molecule to partition between liquid and vapor phases**
  - **Henry's law is used in relatively dilute systems (Ethanol vs. aroma compounds)**
    - 46- 49 M H<sub>2</sub>O or 2-2.8 M Ethanol vs. mM, μM, nM Aroma Compounds
  - **Vapor-liquid equilibrium data are represented in terms of K values**
  - **K value is vapor liquid distribution ratio**
  - **$K = \frac{y_i}{x_i}$**
- **Can be really complex of course:**
  - **Influenced by Chemical Equilibrium**
  - **Temperature (of course)**
  - **Ionic Strength (more salt tends decrease solubility of gases)**
  - **Solvent mixtures (EtOH + Water)!!!**
  - **Non-ideal solutions (sucrose)**

# Research Ongoing

- **GC-QTOF (untargeted & target ) characterization of samples and standards**
- **Understand which compounds and how matrix of aroma compounds influence what we perceive**
- **Understand the relationship between the composition of our standards created for panelists and how they relate to wine composition**

# Take Away Messages

- Results suggest aroma/flavor partitioning into H<sub>2</sub>O/EtOH is largely driving aroma profile in wine.
- Aroma profile in high ethanol wine is characterized by riper characteristics and greater viscosity (lower density).
- Riper fruit yields more color and tannins
  - More saturated color and more astringency.
- Disclaimer: These results may not apply to other cultivars and regions

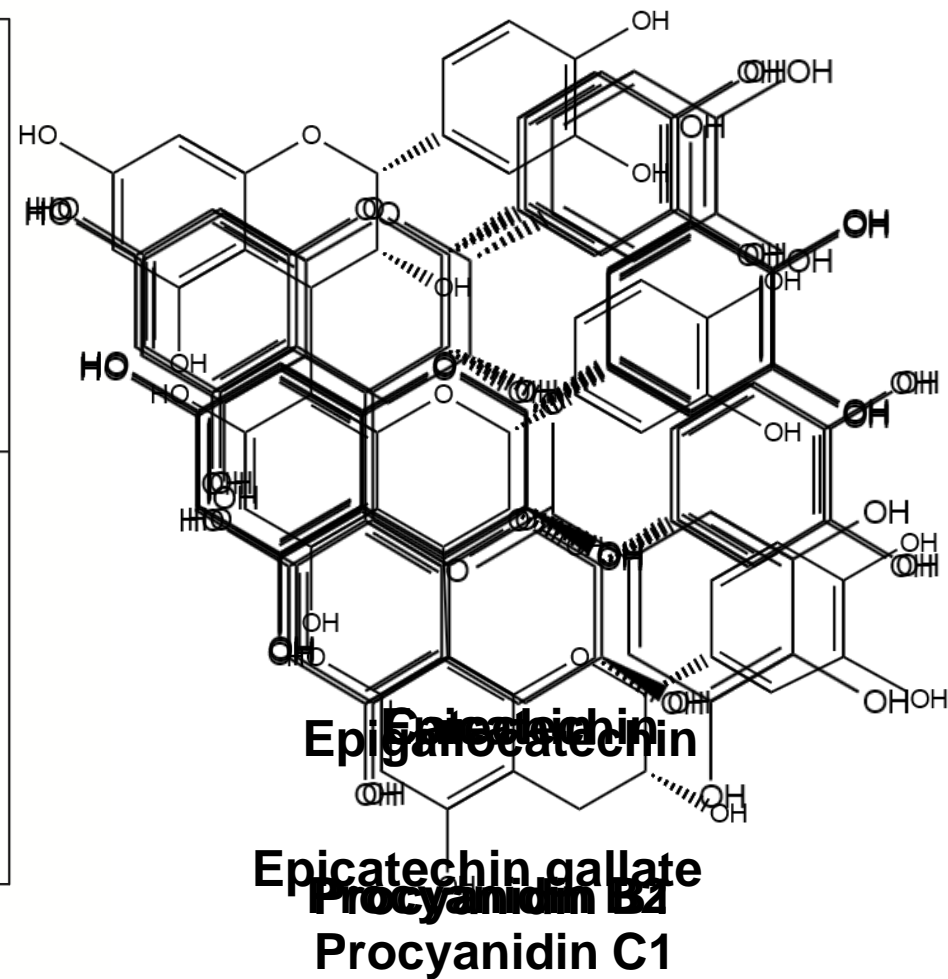
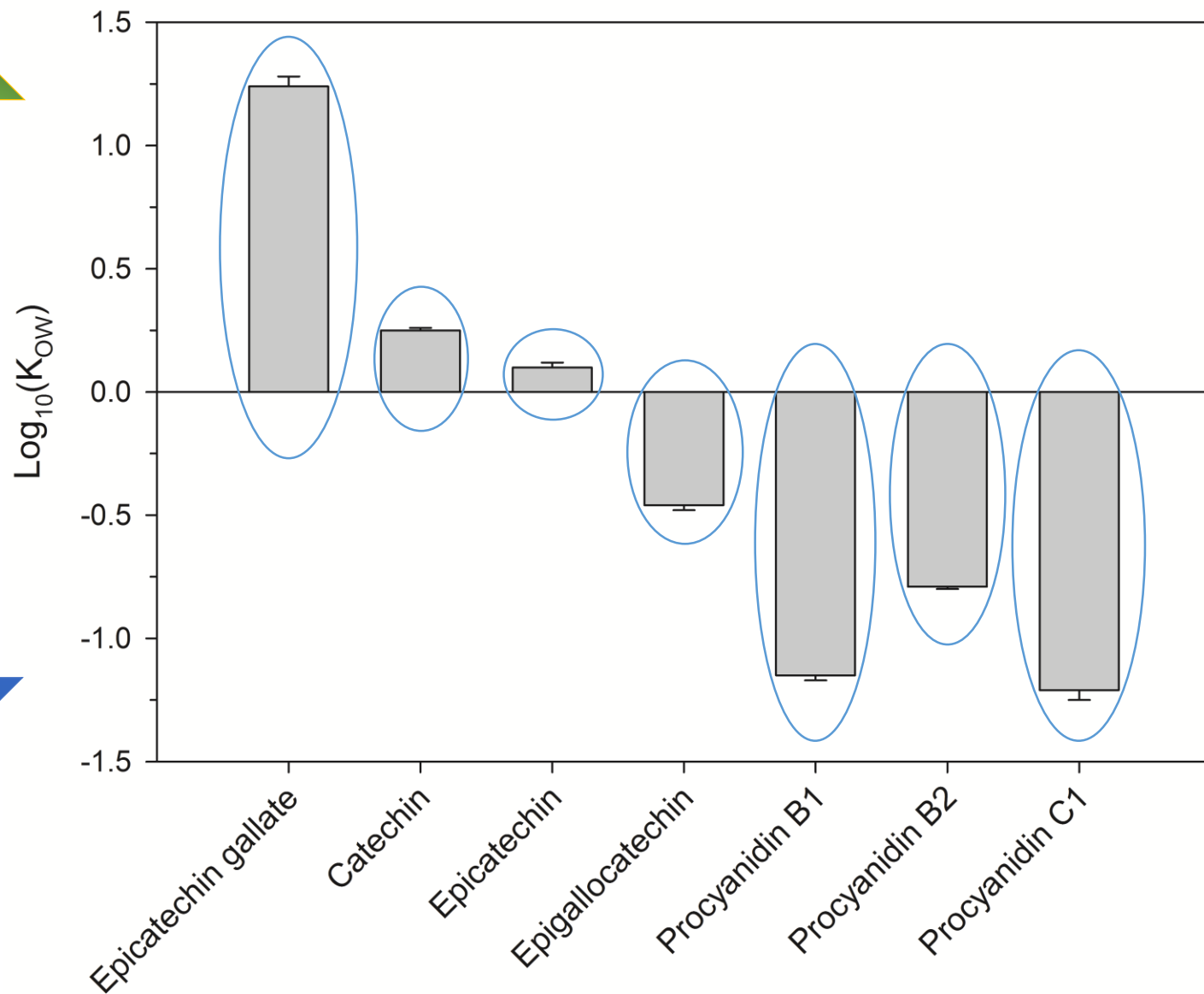
# Acknowledgements

- **Bree Boskov Oregon Wine Board, Mark Chien, James Osborne OSU**
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  - **Wine Advisory Committee, the Washington Wine Commission, the Washington Grape and Wine Research Program, and the WSU Agricultural Research Center**
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- **University of Aukland: Emma Sherman, Dr. David Greenwood, Dr. Silas Villas Boas**

# **Change Gears!!!! Phenolic Hydrophobicity**

- **Hydrophobicity is a measure of how much a compound will dissolve/react with water**
  - **Hydro-water**
  - **Phobia-fear**
- **Hydrophobicity of tannins relates to the strength of the protein-tannin complex**
  - **Number of hydrogen bonds vs hydrophobic interactions**
- **As a tannin polymer gets larger, it also gets:**
  - **More hydrophilic**
  - **More efficient at precipitating protein**
    - **More astringent**

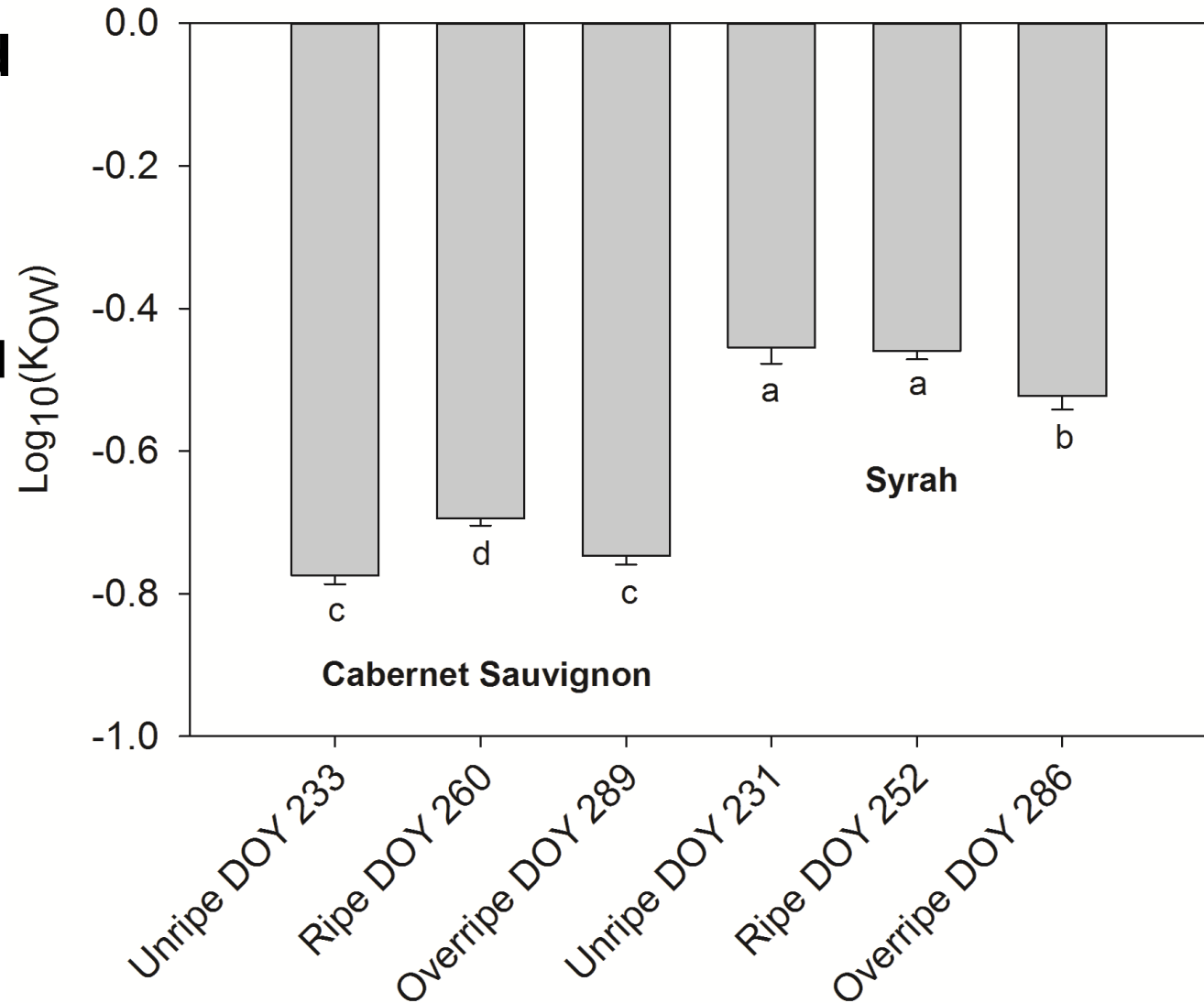
# Phenolic Standards



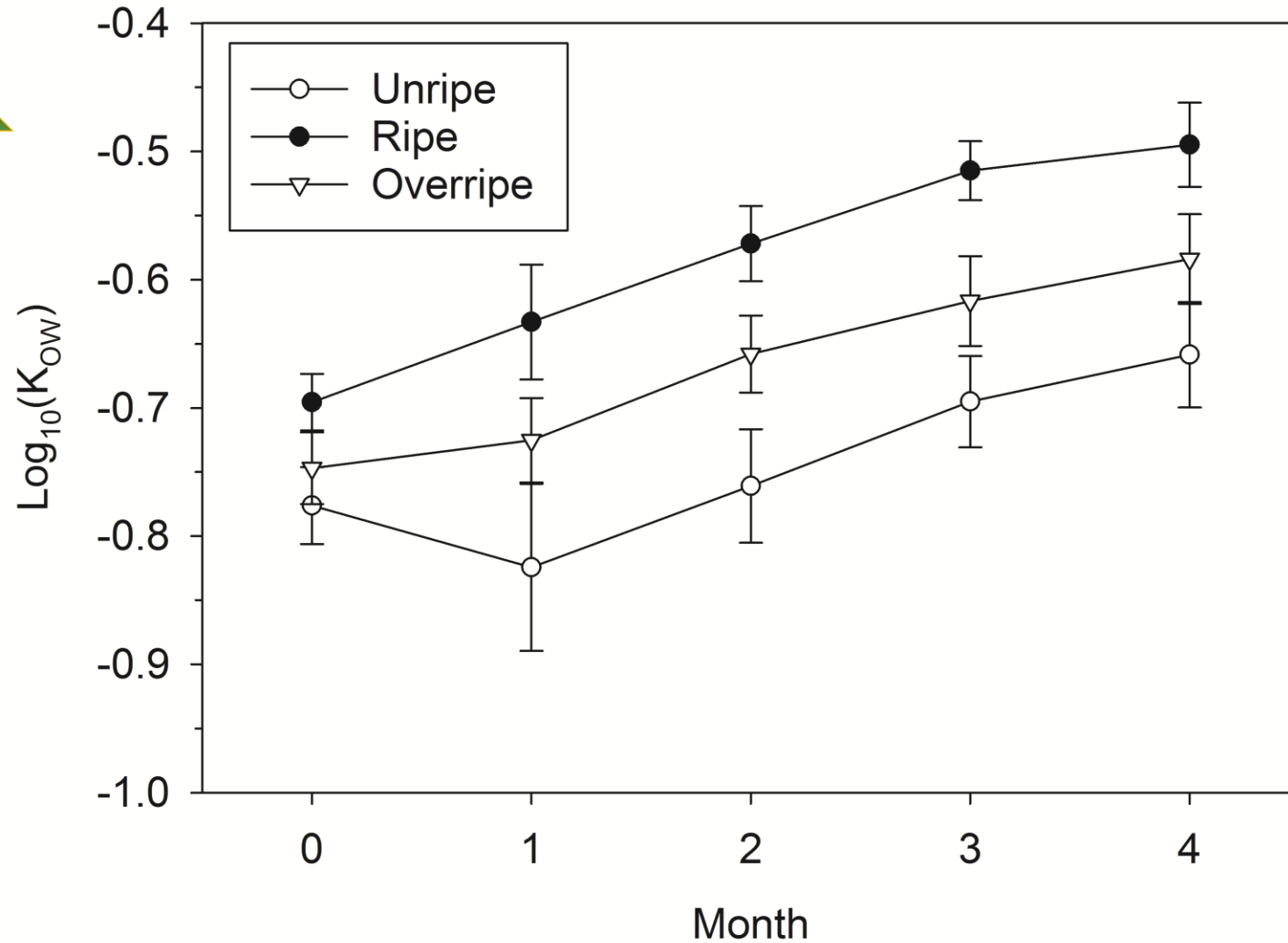


# Wine Phenolic Hydrophobicity

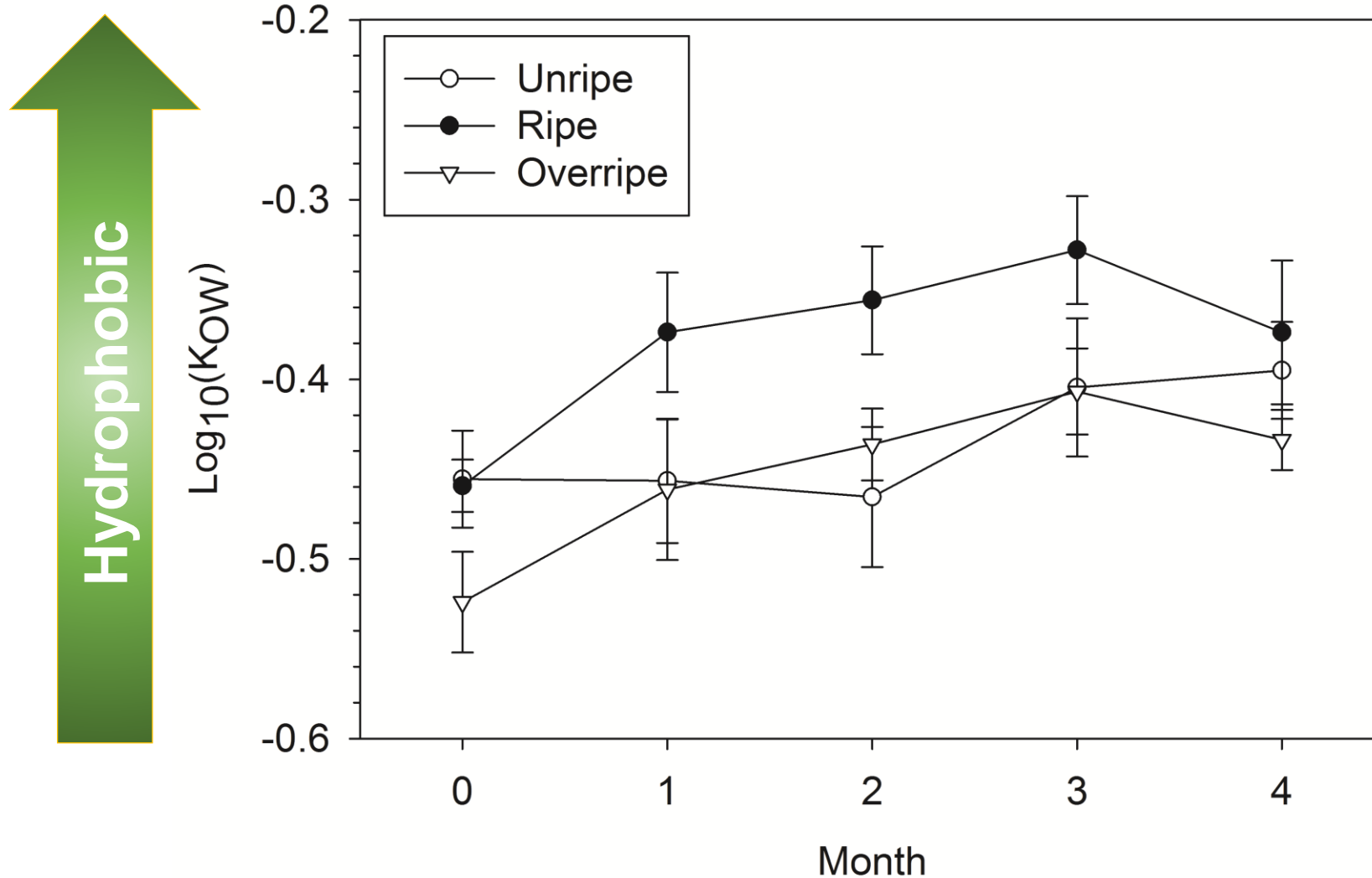
- **Dependent on varietal and berry maturity**
- **Independent of wine alcohol content**
  - **Alcohol content increased tannin concentration but did not change tannin composition**
  - **Anthocyanin content also independent of alcohol**
    - **But increasing anthocyanin would make hydrophobicity decrease**



# Hydrophobicity Over Time-CS



# Hydrophobicity Over Time-SY



# What is happening over time?

- To increase  $K_{ow}$ , either:

$$\uparrow K_{ow} = \frac{\text{Area octanol phase}}{\text{Area water phase}} \begin{matrix} \uparrow \\ \downarrow \end{matrix}$$

- Area of octanol phase is not increasing, but area of water phase is decreasing
  - Losing hydrophilic compounds, not gaining hydrophobic ones
  - Tannin polymers NOT getting shorter

# Conclusions

- **Wine hydrophobicity suggests tannin polymers are relatively small**
- **Phenolic hydrophobicity is dependent on fruit maturity, not alcohol content**
  - **Tannin concentration is not changing with maturity but tannin composition/structure is changing**
- **Phenolic hydrophobicity increases over time**
  - **Due to structural transformations and losses of anthocyanin**
    - **Anthocyanin retention 40-60% over 4 months**
  - **Probably NOT due to shortening of tannin polymers over time**